Krystyna Truszkowska, Cecylia Wesołowska\*

# Influence of preparation conditions of gadolinium oxide films on their optical properties and structure

Optical properties of vacuum deposited films of gadolinium oxide have been studied in the infrared region from 0.6 to 50  $\mu$ m. The refractive index of the films has been determined in the wavelength of 0.6-5  $\mu$ m. The Gd<sub>2</sub>O<sub>3</sub> films are highly transparent in the spectral region of 0.6 to 20  $\mu$ m with the exception of two absorption bands at 3 and 7  $\mu$ m. Interpretation of these bands is given.

## 1. Introduction

The investigations on optical properties of  $Gd_2O_3$ films, conducted earlier, have indicated that these films being highly transparent within a wide spectrum, can be utilized in multilayer stacks. Further investigations, which were undertaken, concern the influence of the deposition parameters, like gas pressure and substrate temperature, on the optical properties and structure of the films, as well as the interpretation of absorption bands in the infrared region.

## 2. Experimental

The films of gadolinium oxide were evaporated by electron bombardment of the  $Gd_2O_3$  powder compressed into pills and pure Gd metal in an oxygen atmosphere at  $2 \times 10^{-5}$  Tr. The films were condensed onto amorphous substrates (borosilicate glass, fused quartz) and crystalline substrates (CaF<sub>2</sub>, KRS-5). The preparation of the films have already been reported in full details [1].

The transmittance was measured by using the following spectrophotometers: VSU2-P, Cary 14, Perkin Elmer 621. The reflectance was recorded on a special reflectance attachment constructed in our laboratory. The film thickness was measured by multiple beam interference (Tolansky's method).

## 3. Results and discussion

The refractive and absorption indices of the  $Gd_2O_3$  films have been determined from measurements, at normal incidence, of reflectance and transmittance. The influence of the  $Gd_2O_3$  evaporation condi-

tions on the refractive index *n* is shown in fig. 1. The curve *a* is the average dispersion curve from several different  $Gd_2O_3$  films deposited onto unheated  $CaF_2$  substrates at  $2 \times 10^{-6}$  Tr. The values of the index are lower than those reported by HASS [2]. Evapo-

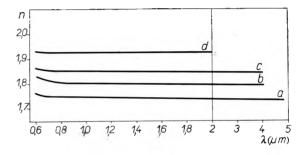


Fig. 1. Refractive index of evaporated  $Gd_2O_3$  films a - films deposited onto an unheated  $CaF_2$  substrate, b - filmsproduced by evaporating a pure Gd metal in oxygen atmosphere, c - films evaporated onto an unheated  $CaF_2$  substrate at  $5 \times 10^{-5}$  Tr of oxygen, d - film on a quartz substrate heated to 200°C. The scale changes at 2  $\mu$ m

ration in oxygen at a pressure of  $5 \times 10^{-5}$  Tr allowed to produce films with a refractive index higher by 5% (curve c). Gadolinium oxide films, prepared by evaporating pure Gd metal at oxygen pressure of  $2 \times 10^{-5}$ Tr onto an unheated substrate, have the middle values of rafractive index. Gd<sub>2</sub>O<sub>3</sub> films showed a very marked dependence of the optical constants on substrate temperature. The index of refraction of the films condensed onto a quartz substrate at 200°C increased from 1.78 to 1.92 (curve d). Such high value of *n* can be obtained by baking the films deposited on unheated substrates in air at 400°C for 5 h after evaporation. The refractive index of all the films has a small dispersion in the studied wavelength range.

The increase of the refractive index of  $Gd_2O_3$ films is connected with the ordering of its structure and an increase of density packing. X-ray diffraction study of the films showed the films formed at room temperature were amorphous in nature, whereas those deposited at 200°C were polycrystalline with a preferred [100] orientation. Annealing of

<sup>\*</sup> The authors are with Institute of Physics, Technical University of Wrocław, Wybrzeże Wyspiańskiego 27, 50-370 Wrocław, Poland

amorphous films on air at 400 °C caused their partial crystallization. Figure 2 shows typical diffraction patterns of  $Gd_2O_3$  powder used in evaporation (*a*), of the film deposited onto a heated substrate (*b*) and of the film baked in air (*c*).

In the investigated infrared 2-50  $\mu$ m region the Gd<sub>2</sub>O<sub>3</sub> films — regardless of the evaporation technique and the kind of substrate — have absorption bands localized at 3  $\mu$ m, 7  $\mu$ m and 25  $\mu$ m (fig. 3). In the case of films thicker than 0.7  $\mu$ m a very weak

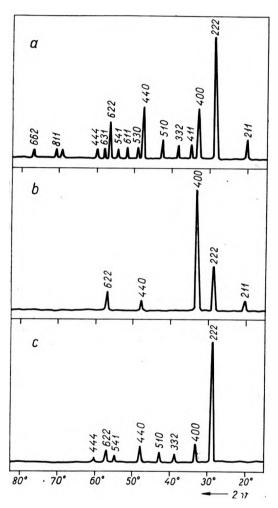


Fig. 2. X-ray diffraction pattern of  $Gd_2O_3$  powder (a), film of  $Gd_2O_3$  on a heated substrate (b), film baked in air (c)

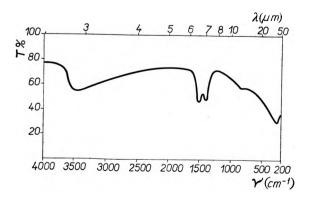


Fig. 3. I. r. transmittance of  $1.0 \,\mu m$  thick film on an unheated KRS-5 substrate

#### Influence of preparation ...

absorption band appears at 11.7  $\mu$ m. We have determined the absorption coefficient *a* at the maximum of these bands: for  $\lambda = 2.88 \ \mu$ m,  $a = 1 \times 10^3 \ \text{cm}^{-1}$ , for  $\lambda = 6.66 \ \mu$ m and 7.14  $\mu$ m  $a = 5 \times 10^3 \ \text{cm}^{-1}$ , for  $\lambda = 25 \ \mu$ m  $a = 7 \times 10^3 \ \text{cm}^{-1}$ . On the basis of comparison of the transmission spectrum of Gd<sub>2</sub>O<sub>3</sub> powder used in our experiment with the spectrum of the films, one may conclude that the absorption band at wavelength 25  $\mu$ m concerns the centre of the lattice vibration of gadolinium oxide [3].

The rare earth oxides being hygroscopic, one would expect that some of the aforementioned bands are due to water incorporated into films. In order to see if this expectation was valid the films were enclosed in an exsiccator with a deuterium oxide for one month. In the transmission spectrum of the films after deuteration an additional absorption band appears with a maximum at 3.9  $\mu$ m. This izotopic shift of the water band is illustrated by dashed line in fig. 4. The experiment confirmed that absorption band at about 3  $\mu$ m is caused by that O-H stretching mode. Fig. 5 represents the transmittance of Gd<sub>2</sub>O<sub>3</sub> film

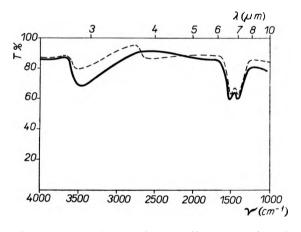


Fig. 4. I.r. transmittance of  $Gd_2O_3$  film on an unheated  $CaF_2$  substrate. Dashed line: the same film after deuteration

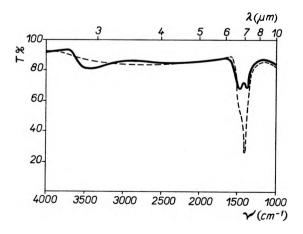


Fig. 5. I.r. transmittance of  $Gd_2O_3$  film on an unheated  $CaF_2$  substrate before being baked in air (continous line) and the same film after being baked (dashed line)

#### K. Truszkowska, C. Wesołowska

on the CaF<sub>2</sub> substrate (continuous line) and the same film after being baked in air (dashed line). The disappearance of the 3  $\mu$ m band is the second proof for the origin of this band. The deformation and the great depth of the double band in the 7  $\mu$ m region are visible after baking treatment.

In order to ascertain that the absorption bands, occurring at  $3 \mu m$  and  $7 \mu m$ , are characteristic of Gd<sub>2</sub>O<sub>3</sub> films only, optical properties of other rare earth oxide films: CeO<sub>2</sub>, Pr<sub>6</sub>O<sub>11</sub>, Nd<sub>2</sub>O<sub>3</sub>, Sm<sub>2</sub>O<sub>3</sub>, Er<sub>2</sub>O<sub>3</sub>, Yb<sub>2</sub>O<sub>3</sub> and Y<sub>2</sub>O<sub>3</sub> have been investigated in the wavelength range of 2–10  $\mu$ m. All the films examined have the absorption bands localized at the same wavelength ranges as Gd<sub>2</sub>O<sub>3</sub> films. The group of lanthanides is highly reactive towards atmospheric gases, such as hydrogen, water, nitrogen and carbondioxide [4]. In the case of rare earth oxides it is possible to form the carbonates Me<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub> or Me(HCO<sub>3</sub>)<sub>3</sub>.

To explain the absorption band at  $7 \mu m$  several experiments have been made with  $Gd_2O_3$  powder, including a thermogravimetric analysis at temperature ranging within 20–950°C, the measurement of infrared transmittance of the powder baked at 400°C and 900°C, and Raman spectrum. From these investigations and the literature data related to the vibration of the  $CO_3^-$  ion in carbonate compounds [5] it may be concluded that at 7  $\mu m$  the double absorption band is due to the stretching vibration of the  $CO_3^-$  ion and at 11.7  $\mu m$  to the bending of the same ion.

## 4. Conclusions

The refractive index of  $Gd_2O_3$  films depends on evaporation condition. Its value increases with the increasing substrate temperature. This appears to be associated with the crystal growth process of the films.

#### Influence of preparation ...

Low values of  $\alpha$  for the maximum infrared bands allow to state that the evaporated  $Gd_2O_3$  films have a wide region of high transparency, and that they find applications in optics.

\*

The authors wish to express their thanks to Dr T. Głowiak for the X-ray analysis and to Dr J. Baran for his helpful suggestions in the experimental work.

## Влияние условий приготовления плёнок из окиси гадолиния на их оптические свойства и структуру

Оптические свойства плёнок окиси гадолиния, полученных путём возгонки в вакууме, исследовались в области инфракрасной части спектра от 0,6 до 50  $\mu$ м. Определили коэффициент преломления этих плёнок в диапазоне длины волны от 0,6 до 5  $\mu$ м. Плёнки Gd<sub>2</sub>O<sub>3</sub> являются очень прозрачными в пределах от 0,6 до 20  $\mu$ м за исключением полос поглащения при 3 и 7  $\mu$ м. Приводится интерпретация этих полос.

## References

- TRUSZKOWSKA K., WESOŁOWSKA C., Thin Solid Films, 34, 1976, p. 391.
- [2] HASS G., RAMSEY J. B., THUN R., J. Opt. Soc. Am. 49, 1959, p. 116.
- [3] MCDEVITT N. T., DAVIDSON A. D., J. Opt. Soc. Am. 56, 1966, p. 636.
- [4] GASGNIER M., GHYS J., SCHIFFMACHER G., LA BLANCHE-TAIS Ch. H., CARO P. E., J. Less Common Metals 34, 1974, p. 131.
- [5] NAKAMOTO K., Infrared Spectra of Inorganic and Coordination Compounds, 2d ed., John Wiley & Sons, New York 1970.

Received November 19, 1976